## VII-F Synthesis of Transition-Metal Chalcogenido Complexes and Their Cluster-Forming Reactions

Transition-metal chalcogenido aggregates are of well-documented importance in biological systems and industrial processes such as hydrodesulfurization. A wide variety of metal chalcogenido clusters have been synthesized, in which the tetrathiometalato anions have been widely used as a building block. In this project, we are focusing on preparation of chalcogenido/chalcogenolato complexes as a precursor for cluster syntheses and their cluster-forming reactions.

## VII-F-1 Synthesis of Bis{(2-dimthylphenylphosphino)ethane-1-thiolato}bis(tertbutylthiolato)molybdenum(IV) and Its Cluster-Forming Reactions with FeCl<sub>2</sub> and CuBr

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The Mo(IV) complex  $Mo(dmsp)(S^{t}Bu)_{2}$  (1) was readily prepared by the reaction of Mo(S<sup>t</sup>Bu)<sub>4</sub> with 2 equiv of HSCH<sub>2</sub>CH<sub>2</sub>PMe<sub>2</sub> (Hdmsp). The X-ray analysis of 1 reveals a distorted octahedral geometry with a cisdisposition of two <sup>t</sup>BuS ligands. Treatment of 1 with FeCl<sub>2</sub> and CuBr led to the formation of heterometallic clusters, [Mo(O)(dmsp)<sub>2</sub>]<sub>2</sub>FeCl<sub>2</sub> (2) and [MoBr(dmsp)<sub>2</sub>- $(\mu_3-S)Cu_2]_2(\mu_2-S^tBu)_2$  (3), respectively. The oxo ligand in 2 is most probably derived from adventitious  $H_2O$ contained in hygroscopic FeCl<sub>2</sub>. In the structure of **2**, an FeCl<sub>2</sub> unit bridges two square-pyramidal Mo(O)(dmsp)<sub>2</sub> fragments through interactions between iron and sulfur atoms of dmsp. The formation of 3 involves C-S bond cleavage of one 'BuS ligand of 1 and rearrangement of ligands between the Mo and Cu sites, resulting in the structure consisting of two MoCu2BrS(dmsp)2 units and two <sup>t</sup>BuS bridges.



Figure 1. Structure of 3.

VII-F-2 Synthesis and Structure of a Triply-Fused Incomplete-Cubane Cluster [{( $\eta^5$ -C<sub>5</sub>Me<sub>5</sub>)-WS<sub>3</sub>}<sub>3</sub>Cu<sub>7</sub>(MeCN)<sub>9</sub>](PF<sub>6</sub>)<sub>4</sub> and a 2D Polymer [( $\eta^5$ -C<sub>5</sub>Me<sub>5</sub>)WS<sub>3</sub>Cu<sub>3</sub>(MeCN)(pz)]PF<sub>6</sub> (pz = pyrazine)

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The reaction of (PPh<sub>4</sub>)[Cp\*W(S)<sub>3</sub>] with 3 equiv of [Cu(MeCN)<sub>4</sub>](PF<sub>6</sub>) in MeCN yielded a triply-fused incomplete-cubane cluster [{( $\eta^{5}$ -C<sub>5</sub>Me<sub>5</sub>)WS<sub>3</sub>}<sub>3</sub>Cu<sub>7</sub>-(MeCN)<sub>9</sub>](PF<sub>6</sub>)<sub>4</sub> (**1**). Furthermore, we constructed a 2D polymeric structure [( $\eta^{5}$ -C<sub>5</sub>Me<sub>5</sub>)WS<sub>3</sub>Cu<sub>3</sub>(MeCN)(pz)]-PF<sub>6</sub> (**2**) by treating **1** with pyrazine in the presence of LiCl. The W<sub>3</sub>S<sub>9</sub>Cu<sub>7</sub> framework of **1** is broken during the reaction with LiCl and pyrazine, providing a WS<sub>3</sub>Cu<sub>3</sub> incomplete-cubane cluster as a building block of the stacked sheet structure of **2**.



Figure 1. Structure of 1.

## VII-F-3 Synthesis and Structures of the Halfsandwich W(VI) Triselenido and W(II) Selenolato Complexes

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The reaction of Cp\*WCl<sub>4</sub> with LiSe<sup>t</sup>Bu at room temperature gave rise to a mixture of syn- and anti- $Cp*_2W_2(\mu-Se)_2(Se)_2$  (1), in which C–Se bond cleavage took place. When the similar reaction was carried out in the presence of <sup>t</sup>BuNC, the W(II) selenolato complex  $Cp*W(Se^tBu)(CN^tBu)_3$  (2) was formed. The structure of 2 was confirmed by X-ray analysis. While two of the isocyanides are nearly linear  $[C-N-C = 175.8(8)^{\circ}]$ . 164.3(7)°], the other contains essentially an sp<sup>2</sup>-type nitrogen atom [N–C–N, 128.7(6)°] amongst the smallest of the known bent isocyanides (122-156°). In another experiment, a freshly prepared Cp\*WCl<sub>4</sub>/LiSe<sup>t</sup>Bu mixture was quickly transferred into Li<sub>2</sub>Se<sub>2</sub> in THF. Cation exchange with PPh<sub>4</sub>Br in CH<sub>3</sub>CN provided  $(PPh_4)[Cp*W(Se)_3]$  (3) concomitant with a mixture of syn- and anti-1. The anion part of 3 has a three-legged piano-stool structure. The average W-Se distance

(2.322 Å) of  $\mathbf{3}$  is similar to that of  $(PPh_4)_2[WSe_4]$ .



**Scheme 1.** Reagents and condititons: i, Li<sub>2</sub>Se<sub>2</sub>, THF; ii, PPh<sub>4</sub>Br, CH<sub>3</sub>CN; iii, 30 min, -78 °C, THF; iv, Bu<sup>t</sup>NC, THF.